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(71) Applicant: APPLIED MATERIALS, INC. [US/US]; 3050 Bowers Avenue, Santa Clara, CA 95054 (US).

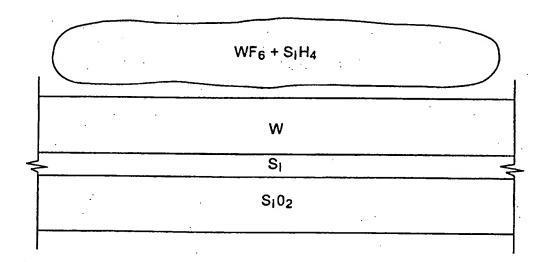
(72) Inventors: MAHAJANI, Maitreyee; 125 Connemara Way #28, Sunnyvale, CA 94087 (US). GHANAYEM, Steve, G.; Apartment 10, 1576 Ontario Drive, Sunnyvale, CA 94087 (US).

(74) Agents: BERNADICOU, Michael, A. et al.; Blakely, Sokoloff, Taylor & Zafman LLP, 7th floor, 12400 Wilshire Boulevard, Los Angeles, CA 90025 (US). (81) Designated States: JP, KR, SG, European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE).

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(54) Title: CVD TUNGSTEN DEPOSITION ON OXIDE SUBSTRATES



(57) Abstract

The present invention discloses a method for depositing a conductor, such as tungsten (W), onto a substrate. The method generally includes depositing an amorphous silicon layer on an oxide substrate and then depositing a conductive film, such as tungsten, thereover. The silicon is believed to provide a nucleation surface for subsequent metal deposition and to provide protection for the substrate from the deleterious effects of the surface reaction between silicon and tungsten hexafluoride.

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CVD TUNGSTEN DEPOSITION ON OXIDE SUBSTRATES

BACKGROUND OF THE INVENTION

Field of the Invention

The present invention relates to chemical vapor deposition, and more particularly, to the chemical bonding between silicon (Si) and tungsten (W) and the deposition of tungsten on a substrate.

Background of the Related Art

In the manufacture of semiconductor electronic devices, ever increasing density of individual devices on substrates has been achieved. Very large scale integration (VLSI) having 10⁵ devices or more per chip have become the norm rather than the exception. The increasing device count (or density) has been accomplished by shrinking the minimum device feature size so that features are less than one micrometer (1µm) in size. Low resistance interconnects using metal films become necessary in such structures to maintain speed performance in the device circuitry. Metals such as aluminum (Al), tungsten (W), copper (Cu) and the like have proven useful as interconnect materials. Tungsten (W) has become preferred in many instances as the first metal deposited on a substrate because of its thermal characteristics, relatively low resistivity, resistance to electro-migration and good step coverage.

Chemical vapor deposition (CVD) of tungsten has largely been employed using tungsten hexafluoride (WF₆) as a precursor in gaseous form. Blanket or selective deposition of a tungsten layer on an oxide surface can be performed by employing a reducing agent, such as hydrogen (H_2), silane (SiH₄) or a mixture of these with the WF₆.

However, tungsten typically does not adhere well to an oxide layer. Therefore, a "glue" layer, typically a nitride of titanium or tantalum, is first deposited on the oxide layer and then the tungsten is deposited thereover as shown in Figure 1. This allows the WF₆ to be reduced by the "glue" layer to provide good adherence to the substrate being processed. The "glue" layer is typically deposited in a separate chamber, such as a CVD or PVD chamber, prior to the deposition of the W layer. Thus, an extra step in the manufacturing process is required to deposit this "glue" layer. Each additional

process step increases the cost of manufacturing and may expose the substrate to contamination during its transfer between chambers.

Therefore, there remains a need for a process to deposit a tungsten film on a substrate which demonstrates good adherence to the substrate.

SUMMARY OF THE INVENTION

According to the techniques of the present invention generally, a single CVD chamber is employed to deposit W onto an oxide substrate without the use of a traditional "glue" layer. An amorphous layer of silicon (Si) is preferably deposited in situ in the CVD chamber prior to the W nucleation and deposition steps to provide good adherence of the W to the substrate. A plasma is struck in the CVD chamber over the substrate to be processed using a silicon source gas, such as silane, and an inert gas, such as argon (Ar). A thin amorphous Si layer is thereby formed over the oxide layer on the substrate. The thin amorphous Si layer is then able to provide a reducing surface on which the WF6 gas, or other W precursor source, can nucleate to initiate and then continue the W deposition process. Good adhesion to the substrate is achieved in this manner at the SiO₂/Si/W interface (comparable to that using a TiN "glue" layer). During the deposition process, it is believed that the WF₆ is reduced to the tungsten metal and the silicon is oxidized to the volatile SiF4. This reaction is believed to improve the adhesion of the W to the oxide layer through the amorphous silicon layer. Additionally, it is believed that the WF6 is also reduced by SiH4 and H2 to achieve continued deposition of W. The amorphous Si layer also serves to protect the substrate form the corrosive effects of the WF6. Further, an unexpectedly good result of lowering the overall W film stress from that typically achieved using a traditional "glue" layer.

BRIEF DESCRIPTION OF THE DRAWINGS

So that the manner in which the above recited features, advantages and objects of the present invention are attained and can be understood in detail, a more particular description of the invention, briefly summarized above, may be had by reference to the embodiments thereof which are illustrated in the appended drawings.

It is to be noted, however, that the appended drawings illustrate only typical embodiments of this invention and are therefore not to be considered limiting of its scope, for the invention may admit to other equally effective embodiments.

Figure 1 is a schematic cross sectional view of a substrate having a Tungsten (W) layer deposited over a TiN layer according to the prior art.

Figure 2 is a schematic cross sectional view of a typical CVD substrate processing chamber used to employ the techniques of the invention.

Figure 3 is a schematic cross sectional view showing the deposition of a thin amorphous Si layer over a substrate according to the invention.

Figure 4 is a schematic cross sectional view showing the deposition of tungsten (W) over the thin Si layer of Figure 3 according to the invention.

DETAILED DESCRIPTION OF THE PREFERRED EMBODIMENT

Referring initially to Figure 2, a schematic cross sectional view of a typical CVD processing chamber of the type used to perform techniques of the invention is shown. An example of such a chamber is WxZ chamber available from Applied Materials, Inc. in Santa Clara, California. The CVD chamber 10 generally includes a chamber body 20 which defines a processing region 22 bounded at an upper limit by a gas distribution assembly 24 and at a lower limit by a substrate support member 26. A vacuum system 21 connected at a lower end of the chamber body 20 maintains an operating pressure in the process region 22 by exhausting the process region 22 via a port 23. The gas distribution assembly 24 is disposed on a lid 27 and comprises a top mounted gas feedthrough 28 coupled to a base plate 30 at a lower end, a faceplate 32 having apertures 33 formed therein, and a perforated blocker plate 34 disposed between the faceplate 32 and the base plate 30. A conduit 35 provides a gas pathway through the base plate 30. Quick disconnect hoses 36 are connected to the gas feedthrough 28 to supply a heating fluid to an annular fluid passageway 38 formed in the base plate 30. Inlet/outlet channels 40 provide a pathway for delivering the fluid to the annular passageway. A cover 39 mounted to the lid 27 shields the gas delivery system 24. Seals, such as o-ring seals 37, maintain the vacuum integrity of chamber 30.

Gas sources 41 are connected to the chamber bottom to supply process gases and carrier gases to the gas distribution assembly 24. A mixing chamber 43 may be

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located upstream from the processing region 22 to mix various gases prior their delivery into the chamber and the rate of gas delivery is regulated by flow controllers 45.

The substrate support member 26 generally comprises a substrate support surface 42 and stem 44 disposed through the chamber bottom. A plurality of grooves 47 formed in the substrate support surface 42 are connected to vacuum pump 49 to provide a backside vacuum to a substrate to hold the substrate during processing. An actuator 46 moves the substrate support member 26 between a lowered loading/unloading position and a raised processing position. Lift pins 48 slidably disposed through the substrate support member 26 are adapted to receive a substrate in the lowered loading/unloading position. The substrate support member 26 contains a resistive heating element (not shown) to heat a substrate during processing. Alternatively, lamps and other known devices can be used to heat the substrate.

The chamber 10 is shown containing a substrate 50 on the substrate support member 26. Figure 2 shows the substrate 50 in a raised processing position. The substrate 50 has its temperature controlled by a temperature control system (not shown) which controls current flow to the resistive heating elements. Processing gases are supplied from the containers 41 to the processing region 22 through the gas distribution assembly 26. The gases are routed through the feedthrough 28 to the blocker plate 34 via the conduit 35. The blocker plate 34 acts as an initial dispersion stage wherein the gases are uniformly distributed. Holes (not shown) in the blocker plate 34 then channel the gases onto an upper surface of the faceplate 32. The gas is delivered into the processing region 22 by the apertures 33 formed in the faceplate 32. A plasma may be generated by delivering a signal, such as an RF signal, to the faceplate 32 while grounding the substrate support member 26. Alternatively, a negative bias may be applied to the substrate 50 by coupling the substrate support member 26 to a signal source. Electric fields inside the processing region may be controlled via an RF power source 52.

The CVD processing chamber 10 of Figure 4 is not intended to represent any particular CVD processing system, but rather to generically illustrate typical chambers which are employed in the industry to coat or deposit various materials on substrates. Other chambers may be used to advantage for the present invention.

Referring to Figure 1, the prior art use of a "glue" layer prior to the deposition of a tungsten layer is shown schematically. As discussed above, the typical tungsten deposition process sequence includes a step of depositing a thin titanium nitride layer (TiN) over the surface of an oxide substrate in a separate chamber. This "glue" layer of TiN then allows deposition of a blanket layer of tungsten (W) thereover as previously discussed in a separate processing chamber.

According to the techniques of the present invention, a thin amorphous silicon layer (Si in Figure 3) is first deposited on the substrate in the processing chamber rather than the titanium nitride "glue" layer. A plasma is struck in an argon/silicon gas mixture introduced into the chamber. This plasma is shown in Figure 3 as a (SiH₄ + Ar) cloud over the substrate. The silicon in the SiH₄ is deposited under the set of preferred temperature, pressure, spacing controlled parameters given below:

Temperature (°C)	425.
Pressure (Torr)	7.5 .
Spacing (Mils)	400
Time (sec)	80
RF power (watts)	175
Ar (sccm)	300
SiH ₄ (sccm)	50

While the preferred value of these parameters is given above, it will be understood by those skilled in the art, that a range of parameters may also be used successfully within the spirit and scope of the invention. For example, the temperature can be in the range of 200 to 550°C, the pressure T in the range of 100mTorr to 15Torr, the spacing of the gas plate from the substrate in the range of 300 to 900 mils, the RF power in the range of 50 to 5000W, the Ar flow in the range of 50 to 1000sccm and the SiH₄ flow rate in the range of 5 to 500sccm. Additionally, other silicon source gases can be used. The time of the process can range from about 10 seconds to about 5 minutes depending on the desired process.

Once the thin Si layer is deposited in the CVD processing chamber, the WF₆ gas is introduced into the chamber 30 and the chamber parameters changed to those

normally used in W deposition using WF₆ gas. Figure 4 shows a substrate having a W layer deposited over the Si layer according to the techniques of the invention. During the deposition process, it is believed that the WF₆ is reduced to the tungsten metal and the silicon is oxidized to the volatile SiF₄. This reaction is believed to improve the adhesion of the W to the oxide layer through the amorphous silicon layer. Additionally, it is believed that the WF₆ is also reduced by SiH₄ and H₂. Generally, a substrate temperature between about 200°C and 400°C can produce an excellent surface with a sheet resistance (R₃) in the range of 10-15 Ω /sq (where R₃ is defined as ρ /t_m for a unit square, where ρ is the resistivity and t_m is the thickness of the conductor). In practice, W layers deposited according to the invention have shown to have adhered to the substrate comparable to that achieved using a traditional "glue" layer prior to the W deposition. Further, by first depositing Si, the substrate is protected from the corrosive effects of WF₆.

Examination and testing of substrates processed according to the invention has shown the unexpectedly good result of a reduction in the overall W film stress from that achieved in W deposition over TiN performed at a lower temperature (in the range of about 360°C to 400°C). A typical film stress in the vicinity of 1.8 x 10¹⁰ dynes/cm² is achieved on a TiN glue coating and W deposition in the temperature range from 360°C to 400°C. Using the amorphous Si deposition technique of the present invention and deposition of W thereover, a film stress of 2x10° dynes/cm² (or a factor of 10 reduction) can be achieved.

In the techniques of the present invention, a single CVD chamber such as that described with respect to Figure 2 can be used to deposit both the Si layer and the W layer. The chamber 10 is adapted to control all relevant parameters, such as the temperature (°C) control, the pressure (T) control, via exhaust 35, the electric field control via RF supply 38, electrode 39 and bias electrode 40, the spacing (of substrate to gas supply, etc.) via lift pins 33 and the type and rate of processing gasses introduced via inlets 34 and distribution controls 34A.

While the foregoing is directed to the preferred embodiment of the present invention, other and further embodiments of the invention may be devised without departing from the basic scope thereof, and the scope thereof is determined by the claims that follow.

CLAIMS:

1. A method for depositing tungsten on a substrate, comprising:

positioning a substrate in a processing chamber;

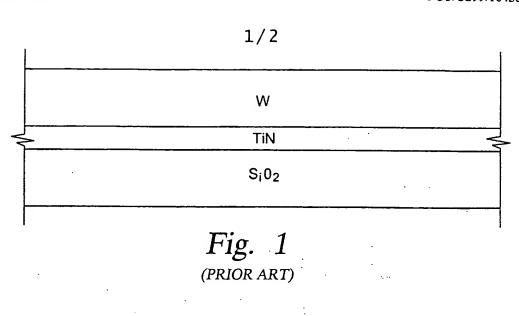
adjusting the operating parameters of said CVD processing chamber to values in the range optimized for depositing a thin silicon (Si) layer, from a first introduced processing gas mixture of a silicon source and an inert gas, onto the oxide surface of said substrate to be processed for a first predetermined time; and

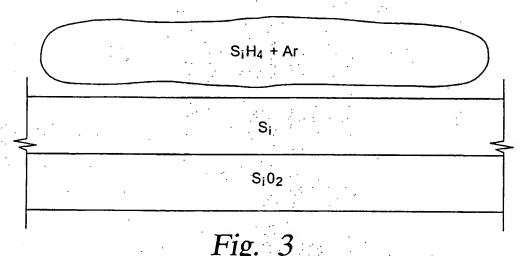
re-adjusting the operating parameters of said CVD processing chamber to those in the range optimized for deposition of tungsten (W) onto said thin Si layer from a second introduced processing gas mixture containing a W source for a second pre-determined time.

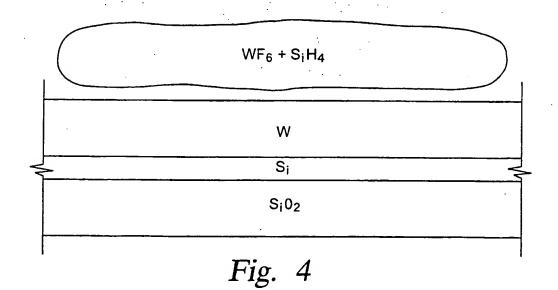
- 2. The method of claim 1 wherein the amorphous silicon layer is deposited over an oxide deposited on the substrate.
- 3. The method of claim 2 wherein the oxide is silicon dioxide (SiO₂).
- 4. The method of claim 1 wherein the silicon source is silane (SiH₄) and the inert gas is argon (Ar).
- 5. The method of claim 1 wherein the tungsten source is hexafluoride (WF₆).
- 6. The method of claim 5 wherein the CVD reaction in the second deposition step is reduction of WF₆ by generally the process

$$2WF_6 + 3Si^* \rightarrow 2W^* + 3SiF_4$$

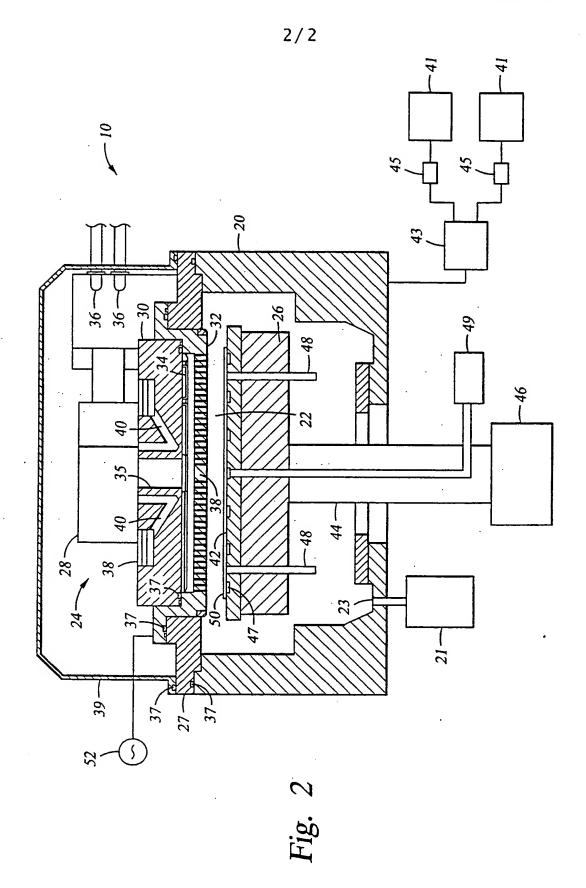
- 7. The method of claim 1 wherein the operating parameters of said CVD processing chamber are optimized for depositing a thin Si layer onto the oxide layer by having the temperature in the range of 350° C to 550° C.
- 8. The method of claim 1 wherein the operating parameters of said CVD processing chamber are optimized for depositing a thin Si layer onto the oxide layer by having the pressure in the range 100mTorr to 15Torr.
- 9. The method of claim 1 wherein the operating parameters of said CVD processing chamber are optimized for depositing a thin Si layer onto the oxide layer by having the first processing gas comprise a mixture of Argon and Silane (Ar + SiH₄) in the ratio range of 10/1 to 5/1.
- 10. The method of claim 1 wherein the operating parameters of said CVD processing chamber are optimized for depositing a thin Si layer onto the oxide layer by having the first processing gas comprise a mixture of Argon and Silane (Ar + SiH₄) in the ratio of 6/1.
- 11. The method of claim 1 wherein the first predetermined time is in the range of 10 seconds to 5 minutes.
- 12. The method of claim 1 wherein the first predetermined time is most preferably in the range of 1 minute to 2 minutes.
- 13. The method of claim 1 wherein the first predetermined time is in the range of 1 minute to 2 minutes.







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INTERNATIONAL SEARCH REPORT

Intr Youal Application No PCT/US 99/16450

A. CLASSIFICATION OF SUBJECT MATTER IPC 7 C23C16/14 C23C C23C16/02 H01L21/285 According to International Patent Classification (IPC) or to both national classification and IPC **B. FIELDS SEARCHED** Minimum documentation searched (classification system followed by classification symbols) C23C H01L Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched Electronic data base consulted during the international search (name of data base and, where practical, search terms used) C. DOCUMENTS CONSIDERED TO BE RELEVANT Category * Citation of document, with indication, where appropriate, of the relevant passages Relevant to claim No. X JP 07 297150 A (NEC CORP) 1-3,5-8,10 November 1995 (1995-11-10) 11-13 column 6, line 25 -column 7, line 15; figure 6 P,X -& US 5 851 581 A (ZENKE MASANOBU) 1-3,5-8, 22 December 1998 (1998-12-22) 11-13 column 5, line 5 - line 59; figure 5 X EP 0 651 436 A (AT & T CORP) 1-3,5,63 May 1995 (1995-05-03) column 2, line 24 - line 57 Further documents are listed in the continuation of box C. Patent family members are listed in annex. Special categories of cited documents: "T" later document published after the international filing date or priority date and not in conflict with the application but "A" document defining the general state of the art which is not cited to understand the principle or theory underlying the considered to be of particular relevance earlier document but published on or after the international "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone filing date "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled "O" document referring to an oral disclosure, use, exhibition or other means document published prior to the international filing date but later than the priority date claimed in the art. "&" document member of the same patent family Date of the actual completion of the international search Date of mailing of the international search report 9 November 1999 22/11/1999 Name and mailing address of the ISA Authorized officer European Patent Office, P.B. 5818 Patentlaan 2 NL – 2280 HV Rijswijk Tel. (+31-70) 340-2040, Tx. 31 651 epo nl, Patterson, A Fax: (+31-70) 340-3016

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Into Jonal Application No PCT/US 99/16450

C.(Continu	tion) DOCUMENTS CONSIDERED TO BE RELEVANT	-	
Category °	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.	
A .	KOW-MING CHANG ET AL: "SiH/sub 4/-WF/sub 6/ gas-phase nucleated tungsten as an adhesion layer in blanket chemical vapor deposition for ultralarge scale integration" JOURNAL OF THE ELECTROCHEMICAL SOCIETY, MARCH 1997, ELECTROCHEM. SOC, USA, vol. 144, no. 3, pages 996-1001, XP002122003 ISSN: 0013-4651 section "Experimental"	1-13	

INTERNATIONAL SEARCH REPORT

information on patent family members

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